

Introspection of chemical transformations in soil through some natural drivers

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Abstract

Soil is regarded as reservoir of essential elements. Living world could not sustain their existence without direct or indirect assistance of soil elements. But the elements of the soil do not always remain in the desired form; instead these elements convert into unwanted irrespective of our consent. Intensive research work on soil revealed some indication that chemical transformation in soil may play some crucial role for changing our environment. The present article focuses on some natural drivers for altering some chemical constituents of soil.

Keywords: *Soil emissions, Green House Gases, Humidity, Temperature, Soil PH, Nutrients*

1. Introduction

Soil is considered to be skin of the earth and is a complex mixture of natural bodies. Since the dawn of civilization it has been serving to ensure the food, energy and handloom supply to human societies. Physical, chemical and biological functions of soil greatly depend on the component present in the soil and organic matter in soil and contribute a lot to soil including nutrient retention, soil structure, and moisture retention, degradation of pollutants, carbon sequestration and soil resilience.

The noticeable report of the Intergovernmental Panel on Climate Change (IPCC) gives a hint that the average elevation of global temperature will increase between 1.1 and 6.4 °C by 2090–2099 as compared to temperature range of 1980–1999 (Solomon *et al.*, 2007). This climate change will greatly influence in environment many locations. The change in climate will not only effects on the environment but also has the potential to threaten many matters including the soil (Brevik., 2012). This threaten will automatically stresses on food security through its influence on soil properties (Brevik and Burgess, 2013). On the other

hand, alteration of soil by the various chemical transformations within the soil elements through the emission of greenhouse gases will greatly impact on the environment. For understanding these effects on both soil and environment by their own influence, climate-soils interaction and how changes in soil will lead to corresponding changes in environmental factors now and again is a great concern. Therefore, this paper will focus on the changes in soil through the chemical transformation of soil constituents by the various natural driving agents including humidity, Temperature, soil PH and availability of nutrients with a particular focus on the soil organic carbon and soil N changes into their oxide and hydrides. The article also highlights the fact that how the change of chemical constituents in soil may alter soil properties and consequent indication of future critical environment.

2. Drivers affecting Chemical transformation in soil:

Soil can be altered by some drivers and it can produce some greenhouse gases (GHG). The most critical factors reported to impact soil GHG generation rates include microbial activity, chemical decay processes, root respiration as well as heterotrophic respiration of soil fauna and fungi influence in production of GHGs in soils (Chapuis-Lardy *et al.*, 2007)., In addition to these factors, soil moisture content (humidity), nutrient availability in soil, soil temperature, soil organic matter and soil aeration, porosity and water, net primary productivity, vegetation type and pH-values are also dominant factors for related emission flux rates (Ludwig *et al.*, 2001, Schlesinger and Andrews, 2000). Some important parameters like Land-cover-related parameters can also be considered as responsible determinant for soil emission. Drivers

can be separated into proximal drivers that influence soil emissions in the direct environment (e.g. local climate, soil type) and distal drivers that effect soil emissions on larger scales (e.g. temperature, humidity); (Robertson, 1989).

2.1. Humidity:

Soil humidity is one of the most significant soil parameter for Soil chemical transformation by emitting GHGs gas from soil because of its controlling power on microbial activity and all associated processes in soil. Nitrifying bacteria reside in soil pores and oxygen is needful for their living. Ludwig *et al.*, 2001 reported that emissions by nitrification even with a maximum at 20% WFPS have been observed by Soils with less water-filled pore space (WFPS) (Ludwig *et al.*, 2001). It was observed that because of inhibited nutrient supply in soil having below 10% WFPS, emissions of Nitric oxide (NO) decrease to some extent (Brümmer *et al.*, 2008). In comparison of Nitrification, yields for NO production found to higher potential than that for N₂O production (Fowler *et al.*, 2009). On the other hand, anaerobic conditions are important requirement for CH₄- and N₂O-producing bacteria. Around 60% WFPS, N₂O production is optimal and found to be lowest when WFPS is below 30% (Gao *et al.*, 2014). It was further investigated the increase of N₂O emissions with an increase of WFPS above 80% (Keller and Reiners, 1994). Soil humidity is directly correlated with CH₄ production for which strictly anaerobic conditions are required (Gao *et al.*, 2014; Smith *et al.*, 2003). Therefore it can be said that under aerobic conditions soils are supposed to be CH₄ sinks (Fiedler *et al.*, 2005). Recently, Wetlands and rice paddies have been identified as strong sources of CH₄ (Dutaur and Verchot, 2007). Due to flooding of rice field, anaerobic fermentation of soil organic matter occurred because of stopping oxygen supply from the atmosphere to the soil. Methane possesses a great power beyond carbon di oxide to influence atmospheric warming. Aridity for Long periods can significantly minimize soil emissions. Then Soils may switch into a sink of N₂O (Goldberg and Gebauer, 2009). Soil moisture also depends on Grain-size distribution. Aerobic condition encourages the emission of gases because Soils having large pores retain less water. (van der Weerden *et al.*, 2010). Soils with unpolished texture give highest NO emission (McTaggart *et al.*, 2002). under anaerobic conditions, production of CH₄ and N₂O observed in soils with dominant fine pores (Dutaur and Verchot, 2007; Gu *et al.*, 2013). During the warm and dry periods, greater amount of CO₂ emissions observed with fine textured soils in comparison with sandy soils (Dilustro *et al.*, 2005). Due to less availability of C and N for soil microbes, Stable soil aggregates (concretions, crusts) conduct to lower soil emissions (Kögel-Knabner *et al.*, 1998).

The pulsing or Birch effect is caused by Precipitation after extended dry periods. (Birch, 1958). Enhancement of emissions within few minutes or hours after the onset of precipitation and come-back to background levels within a small number of days (Sponseller, 2007). The driving factor for the fact is the renewed mineralization and the presence of available easily decomposable material (Borken and Matzner, 2009) for the metabolism of reactivated microbes (Ludwig *et al.*, 2001). The extent of Birch effect diminutions with increase in frequencies of wet-dry cycles (Borken and Matzner, 2009).

To fulfill the demand of food security for growing global population, irrigation is very much important to achieve the goal of sufficient crop production, it incite soil CO₂ emissions in indirect way. (Raich and Schlesinger, 1992) Increase of microbial activities for the rewetted dry soil is associated by the release of air trapped in the soil pores, which is also responsible for increase of CO₂ emission in soil (Rastogi *et al.*, 2002, Sparling, G.P. and Ross, 1988). the main reason for Rewetting of dry soils are irrigation or rainfall which increases CO₂ emission by increasing microbial activity, carbon mineralization and respiration (Sparling, G.P. and Ross, 1988, Calderon and Jackson, 2002) Soil moisture content greatly affects soil respiration and as the result of soil CO₂ emission (Sainju *et al.*, 2012). According to the observation of Ding *et al.*, finding of peak soil CO₂ emission is at 70% soil moisture level (Ding *et al.*, 2007); whereas above 70% soil moisture level, CO₂ emission declined sharply. Rochette *et al* noticed that the rise in soil CO₂ emissions with increasing soil moisture content was accompanied with rising of microbial population and activity (Rochette *et al.*, 1991).

It is ultimate from the observation that Increased in soil moisture is known to responsible to elevated soil emissions of N₂O (Castellano *et al.*, 2010, Bateman and Baggs, 2005).

2.2. Temperature:

Soil temperature can be considered as one of the main abiotic driving factors of soil chemical transformation and soil emission. Soil temperature is also known to explain the variations of GHG emissions from soils (Fang, and Moncrieff, 2001, Toth *et al.*, 2018, Franzluebbers *et al.*, 1995, Hu *et al.*, 2018). Schindlbacher *et al.*, observed that combination of soil temperature and soil moisture may explain a significant part of variations in NO (up to 74%) and N₂O (up to 86%) emissions for particular soils (Schindlbacher *et al.*, 2004). Horel *et al.* 2018 found the significant differences between CO₂ emission values and influence of seasonal effects on soil respiration, and their reducing effect after plant and fruit maturity. Their study also explained the couple effects of soil moisture and temperature on soil GHG emissions, rather than

these individual environmental factors (Horel *et al.*, 2018).

Soil temperature is one of the major factor prevailing microbial activities. Because nitrification and denitrification both processes have been exhibited as being directed by nitrification and denitrification bacteria, soil temperature could affect both transformation processes through its effects on bacterial activities. Various studies revealed that nitrification process preferred temperature ranging from 25 - 35°C, whereas, it will be suppressed when soil temperature decreased below 5°C or increased above 50°C (Well and Brady, 2017). Furthermore, for denitrification the favourable temperature range falls within 30-67°C (Huang *et al.*, 2000).

It has been observed increasing rate of soil emissions and soil respiration rates occurred significantly due to increase of soil temperature. As a result of which, response of microbial metabolism increased. With Increase in temperature, soil respiration rate increases and additional force of Methane and N₂O emissions leads to diminution of O₂ concentrations in the soil (Butterbach-Bahl *et al.*, 2013). As water is essential as a transport medium for nutrients required by microbes, the positive temperature effect may be overlain by soil water stress, (Fowler *et al.*, 2009). NO and N₂O emissions manifested a positive exponential rapport with soil temperature. (Fangand Moncrieff, 2001; Ludwig *et al.*, 2001; Tang *et al.*, 2003, Schindlbacher *et al.*, 2004, Mazzetto *et al.*, 2014). Brooks *et al.* observed that soil respiration from bacteria down to soil temperatures of -7°C (Brooks *et al.*, 1997). The obedience of gas emissions from soils on temperature can be expressed with the temperature sensitivity factor (Q₁₀). Basically it is the measure of the rate of change in a biological or chemical system according to increase in temperature by 10⁰C (Berglund *et al.*, 2010) and it also increases with soil depth (Tang *et al.*, 2003). It has been observed that with the help of this factor soil respiration can also be expressed. This is the factor by which soil respiration enhances by a 10⁰C increase in temperature (e.g., Kirschbaum, 1995; Van't Hoff, 1898). The said approach is implemented in the case of several models, which generally employ a fixed value of 1.5 (e.g., in the Community Land Model, CLM, (Foereid *et al.*, 2014)) or 2 (e.g., in CASA and TEM (Potter *et al.*, 1993; Raich *et al.*, 1991)), which is employed for all soils and soil moisture levels. Nevertheless, various studies indicate that Q₁₀ is variable, with a range of 1 -12 (Gritsch *et al.*, 2015; Hamdi *et al.*, 2013). From the work of Zhou *et al.* (2009a) it has been explored that on a global scale, those small inaccuracies may result in large errors with regard to Q₁₀ in the estimation of carbon dynamics. it has been observed from the data of review by Raich and Schlesinger (1992) that Q₁₀ is 2.4 with values ranging from 1.3 to larger than 3.3

for soil respiration; The average values are authenticated by recent studies of Hu *et al.* (2015) with the values ranged from 1.7–2.5 and Jiang *et al.* (2015) with an average value of 2.2. The fluxes of CH₄ emission to temperature enhances with a Q₁₀-value of ~ 4 (Dalal and Allen, 2008). Increase in N₂O emissions with temperature up to about 37⁰C; according to that denitrification and mitigate N₂O emissions. From the compiled report it has been observed that reported Q₁₀-values for N₂O ranging from 1.7- 9.3 (Abdalla *et al.*, 2009a).The extensive studies revealed some overlapping in moisture and temperature effects, as the result observation of clear correlation in this regard become very difficult (Fangand Moncrieff, 2001). Temperature is prominent factor for the regulation of freeze–thaw events, driving gas emissions from soils (Holst *et al.*, 2008), and may be attributed for up to 50% of the grand annual N₂O emissions (Groffman *et al.*, 2009). On the other hand, CO₂ emissions in winter are considered insignificant for the annual emission budget due to lesser root respiration in temperate or high polar environments (Groffman *et al.*, 2006). Hence, soil water content has to be intimate to saturation to knock off the O₂ content (Groffman *et al.*, 2009). During freeze–thaw cycles, extra nutrients are liberated for microbial metabolism through the soil particles disaggregation (Christensen and Christensen, 1991). After decomposition, dead organic material (e.g., dead plant roots) oppresses increased microbial soil respiration and emissions of N₂O (Mørkved *et al.*, 2006). It is clear from many observations that winter emissions are most pertinent for the temperate climate zone. Independently, maximum soil emissions result in spring. The Changes of environmental factors like the soil wetting before and after rainfall with elevated air temperature, may also result an increase in emission of N₂O (Davidson, 1992, Adviento-Borbe, *et al.*, 2007). Blackmer *et al.* 1982 observed that soil water and temperature are also instigating factor for diurnal variability in N₂O emissions can affect the measurement outcomes, e.g., measurements in mid-day can comprehensively vary from late afternoon or early morning measurements. Apart from that, authors also found site dependent diurnal patterns having strong patterns one study site and no clear diurnal patterns in another site (Blackmer, *et al.*, 1982). From the several studies, it is believed that the static chamber method may overestimate N₂O fluxes in the early afternoon. Hence, early morning or late afternoon assumed to be the preferred measurement time interval (Blackmer, *et al.*, 1982, Alves *et al.*, 2012, Ryden *et al.*, 1978). However, shaded areas might be less impressible for diurnal patterns (Blackmer, *et al.*, 1982) and during collecting sample it should be taken into consideration. Horel *et al.* observed the clear seasonal effects between temperature and soil CO₂ emissions. During

summer, daily soil temperature range (measured at 15 cm depth) reached 4.8 °C (averaging a 3.0 °C difference), On the Contrary, during spring and fall, the maximum daily changes were 5.1 and 4.2 °C (averaging 2.9 and 1.7 °C), respectively (Horel *et. al.*, 2018). Whereas, Aforesaid changes are more traced in air temperature values, the emission of main soil N₂O drivers are soil temperature together with soil moisture values changes, instead of air temperature (Dobbie *et. al.*, 1999). In contrast, intensive correlations between soil CO₂ emissions and temperature are more likely as longer daylight periods and higher temperatures influence plant growth, and as a result, plant root, heterotrophic, and mycorrhizal respiration, which are the premier biogenic sources of soil CO₂ efflux (Fenn, *et. al.*, 2010, Kuzyakov, 2006).

Fidel *et. al.* analyzed the impact of biochar on soil CO₂ and N₂O emissions from Midwestern agricultural soil in a controlled laboratory incubation and in a field under four cropping systems. They hypothesized that biochar's impact on CO₂ and N₂O emissions will significantly depend on soil moisture, temperature and cropping system. In addition to that biochar may suppress the N₂O emissions by reducing the moisture sensitivity of soil nitrous oxide production and not only that but also biochar amendment will show the similar effect on soil N₂O emissions at the laboratory and field vice versa under specific condition but not all condition (Fidel *et. al.*, 2019). Joseph *et. al.*, 2018 showed a correlation among the soil moisture and soil temperatures with soil respiration (Joseph *et. al.*, 2018). They observed effective and positive correlation of soil respiration with soil temperature in all systems. Their studies showed that the rate of soil respiration altered among various systems and was found to be higher during summer season with respect to winter season. The highest rate of soil respiration among the studied was recorded in system of cardamom and coffee and lowest in rubber. Continuous emission of nitrous oxide (N₂O) and methane (CH₄) gases was not observed in different agroforestry systems.

2.3. Soil pH-values

Soil PH is the key factor for influencing microbial activity. There is a direct impact of pH not only on the microbe or physico-chemical environment of soil. According to the different pH optima of different microbial groups, soil has a great diversification in their population balance, biodiversity and related regards from neutral to alkaline soils. While the biodiversity of soils now a day has become an environmental concern, biochemical processes arbitrated by the populations are of most relevant to crop productivity. Nitrification is one of such process which is obstructed at low pH, but this obstruction can be overcome by humic acids. With increasing pH, Nitrification increases, since the equilibrium between

NH₃ and NO₃⁻ shifts to ammonia (Nugroho *et. al.*, 2007). However, no important correlations between NO and N₂O emissions and pH-value were found (Pilegaard *et. al.*, 2006). Denitrification results NO emissions under acidic soil conditions, whereas alkaline conditions nourish NO emissions caused by nitrification (Remde and Conrad, 1991). It has been observed that acidic soil conditions drive lower soil emissions. The pH optima form methanogenesis (CH₄ production) lies between pH 4 -7 (Dalal and Allen, 2008). Emission of CO₂ was observed to be maximum at neutral pH-values (Cuhel *et. al.*, 2010). Under acidic conditions of soil, N₂O emissions decrease. The mitigation of soil N₂O emission is vital concern for sustainable development in agricultural field considering the effect of N₂O in environment various processes have been developed to be useful in controlling soil N₂O emissions. Use of nitrification inhibitors established itself as one of the significant controlling factor.

2.4. Availability of Nutrients

Availability of Nutrient is foremost to microbial and plant respiratory system. It is well known to us that soil microorganisms lead a major role in the deterioration and recycling of organic matter, leading the activities of mineralization, which results the release of inorganic nutrients owing to the benefit of plant roots. According to the knowledge on plant-microbe interactions, the microbes are living mediators which naturalize the oxidation of carbon and mineralization of organic forms of P and N to inorganic and bioavailable forms. Therefore, it is important to mention that natural content of C and N in soil, along with atmospheric deposition, manure or fertilizer applications also circulate an important role. The emissions of N₂O found to be negative correlation with the C/N-ratio (Pilegaard *et. al.*, 2006), with N₂O emissions found lowest at C/N-ratios ≥ 30 (limited disintegration of organic material) and highest at a C/N-value of 11 (optimum disintegration and humus build-up); Gundersen *et. al.* (2012a,b). In combination with drought and low pH-values, emissions of N₂O can be suppressed significantly at C/N-ratios <20 (Gundersen *et. al.*, 2012a,b). CO₂ and CH₄ emissions correlate positively with the ratio of C/N value (Shi *et. al.*, 2014; Weslien *et. al.*, 2009). Nevertheless, CH₄ production in soils may be suppressed due to the availability of other electron donors such as Mn⁴⁺, Fe³⁺, NO³⁻ and SO₄²⁻ (Achnich *et. al.*, 1995; Dalal and Allen, 2008; Fumoto *et. al.*, 2008; Kögel-Knabner *et. al.*, 2010; Sahrawat, 2004), which is of individual relevance in many sub-tropical soils and specially in rice paddies. Considering carbon not limiting, with increasing of soil N content generally leads to soil respiration in higher scale and net ecosystem exchange (NEE) in greater extent (Niu *et. al.*, 2010; Peng *et. al.*, 2011). N fertilizer application

with limited C availability has restricted influence on soil respiration (Micks *et. al.*, 2004). The N fertilizers application leads to a lesser sensitivity of soil respiration against soil temperature and a greater sensitivity to soil moisture (Peng *et. al.*, 2011). Micks *et. al.*, 2004 studied that application of NO_3^- or NH_4^+ leads to diminish or to no change in the rate of soil respiration of forest soils (Micks *et. al.*, 2004). During long-term experiments of N addition, soil respiration decreases (Bowden *et. al.*, 2004). Under aerobic soil conditions, the application of liquid manure (urea) led to higher N_2O emissions, whereas, under saturated conditions NH_4^+ fertilizers caused higher N_2O emissions (Tenuta and Beauchamp, 2003). For minimizing N_2O emissions from agricultural lands, application rates of fertilizer need to be adapted to plant needs due to the fact that all forms of nitrogen cannot be taken up by plants. Non plant-available N amounts may lead to enhancing emissions of N_2O (McSwiney and Robertson, 2005). CO_2 and N_2O fluxes may be increased by Cover crops after their incorporation and significantly impact on soil respiration during their lifetime (Sanz-Cobena *et. al.*, 2014b). Increased level of N_2O emissions can be prevented using controlled-release fertilizers or denitrification inhibitors (Shoji *et. al.*, 2001); yet, this effect can be oppressed by heavy precipitation events (Venterea *et. al.*, 2012). One of the most important factors is Soil water content for selecting the fertilizer type to resist increased N_2O fluxes (Sanz-Cobena *et. al.*, 2014a). Tillage system may also influence fertilizer applications. Under no-till and conservation tillage using urea, N_2O emissions found to be higher, while use of urea-ammoniumnitrate fertilizers could be observed no differences (Venterea *et. al.*, 2005). But Different observations on the influence of the tillage system was also found where there are some finding of decrement of N_2O emissions with no-till practice happened (Omonode *et. al.*, 2011) and explanation in this regard has been described by lower temperatures of soil (Six *et. al.*, 2002), whereas, others established a positive effect of no-till on N_2O emissions and narrated this with higher microbial activities (Baggs *et. al.*, 2003). N_2O emissions in greater extent occur due to the higher soil moisture during no-till practice. In this connection, it has been observed that under no-till system higher N_2O emissions cannot be balanced by CH_4 -uptake rates and higher C sequestration (Li *et. al.*, 2005; six *et. al.*, 2002). However, no information was included about the beginning point of the no-till practice for said study. From the review by six *et. al.*, 2004 regarding impact of no-till practices on emissions of N_2O , a long term studies has been recorded (six *et. al.*, 2004). They found only higher N_2O emissions during the first ten years after switching from conventional tillage to conserving farming practice with no-till. Soil compaction resulted from the use of farming

machines may lead to decrease soil emission (Mordhorst *et. al.*, 2014). At the forest sites, N deposition reduction led to reduce soil NO emissions. On the contrary, it has been found from other studies that by the reduction N_2O emissions were not affected (Eickenscheidt and Brumme, 2012). Abatenh *et. al.* reviewed on microbial function to mitigate GHG emission and role of microbes to fight climate change and GHG reduction through nutrient recycling. It acts as either generators or users of these gases (Abatenh *et. al.*, 2018).

From the observation of Joseph *et. al.* it is clear that during summer months the relatively high soil temperature ($30\text{-}32^\circ\text{C}$) experienced availability of sufficient soil moisture (18.74 - 20.39 %) Due to which, the decomposition of plant litter might have enhanced in all systems by increasing activity of the micro flora and fauna present in soils, consequently, the rate of soil respiration increases (Joseph *et. al.*, 2018). The soil moisture content less than 10 per cent greatly influence the activity of microorganisms in soil (Reshi *et. al.*, 2007). Soil respiration rate was found to be minimum in both morning and noon hours of December. The reason behind the fact is lower soil temperature ($18.6\text{-}24.34^\circ\text{C}$) which minimizes the microbial activity and decomposition of soil organic matter thus driving to carbon dioxide emission in lesser extent. During summer month, the precipitation received in the study fields contributed to sufficient moisture in soil. Precipitation is considered to be a significant factor influencing soil respiration variability in soil and there was an effective increase of the soil respiration rate instantly after the rainfall in summer (Law *et. al.*, 2001).

3. Conclusions

Soil acts not only a sink of reactive carbon, but it handles the global carbon cycle which is directly linked with atmospheric chemistry, ecosystem diversity and radioactive forcing. As per International energy outlook, (EIA 2017), rate of global CO_2 emissions will be increased by 16 per cent between 2015 and 2040. According to the biennial update report (BUR,2016), In 2010, India released 2,136.84 million tons of CO_2 equivalent GHGs and the major source identified in this regard agriculture sector which occupied as 3rd ranking among the emitter of GHGs in the world (Stirling *et. al.*, 2017). The present article revealed the action of some natural drivers causing the subsequent release of soil constituents into their oxides or hydrides. Hence, thinking of soil management should be a desirable practice during the forthcoming decades, as the GHGs emitted from soils sinking the atmosphere should be reduced to minimize adverse effect on the environment. The present work will boost the new researcher to find out the new way, techniques, methodologies and process to control our

environmental hazards and save the global environment.

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