

# Physico-chemical Characterization of Ground Water with Reference to Water Quality Index and Their Seasonal Variation in Vicinity of Thermal Power Plant at Yamuna Nagar, Haryana

Atul K. Ahuja<sup>1</sup>, Prem Singh<sup>2</sup> and Varinderjit Singh<sup>3</sup>

<sup>1</sup>Dept. of Physics, I.B. College, Panipat, Haryana, India.

<sup>2</sup>Dept. of Physics, S.D. College, Ambala Cantt., Haryana, India.

<sup>3</sup>Dept. of Physical Sciences, I.K. Gujral Punjab Technical University, Kapurthala, Punjab, India.

## Abstract

In coal based thermal power plants, the effluents like fly ash, either in dry or slurry form is the major source of contamination of ground water with the toxic elements present in fly ash. This problem is having potential significance in India, as coal used in these power plants has high ash content. In the present research paper, an effort has been made to study the physico-chemical characterization of ground water with reference to water quality index (WQI) in villages around thermal power plant, Yamuna Nagar, Haryana. Water samples from the tube well of 4 villages and a thermal colony surrounding the thermal power plant in Yamuna Nagar, Haryana, were collected for analysis in December 2017 and June 2018. For characterization, various standard methods like inductively coupled plasma emission spectrophotometry (ICPE) and atomic absorption spectrophotometry (AAS), etc have been employed and water quality index is calculated using Tiwari & Mishra method. The standards values of parameters used are incorporated from World Health Organisation (WHO) and Bureau of Indian Standards (BIS). Different Physico-chemical parameters for determining the ground water quality in the vicinity of thermal power plant, Yamuna Nagar are much higher than the standard given by WHO (2017) and BIS (10500:2012) and thus water quality index values calculated from these parameters indicates very poor quality of water and water is not fit for drinking purposes.

**Keywords:** AAS; ICPE; Physico-chemical; WQI; Yamuna Nagar

## 1. Introduction

Throughout the ages, water has been considered pure medium carrying nutritional elements for survival of every form of life on the earth. Water covers about 71% of earth surface. From biological direction, water possesses a lot of different characteristics that are critical for propagation of life. However, in the last few decades, due to accelerated pace of industrialisation and urbanisation, there has been a tremendous decline in the drinking water quality due to its overexploitation and inappropriate disposal of industrial effluents disposal, particularly in urban areas [1]. The quality of drinking water is governed by appropriate values of physico-chemical parameters like Electrical conductivity(EC), pH, turbidity, total dissolved solids etc and certain permissible limits of elemental concentration of toxic trace elements [2]. These has to be in a strict limits, which are governed by certain water standards regulatory authorities like Bureau of Indian standards(BIS), United State environmental protection agency(USEPA), Indian council of medical research (ICMR), World health organization (WHO) etc [3]. These agencies strictly and regularly monitor the water standards. In spite of adhering to the various guidelines issued by these authorities regarding the safe and appropriate disposal of the effluents, most of the industries at state or even central level are not taking proper care for the management of safe disposal of effluents, e.g., fly ash as a by-product produced in tons during coal combustion in the thermal power

plants [4-5]. The Coal fired thermal power plants are producing not only tons of fly ash, but also SO<sub>x</sub>, N<sub>2</sub>, some hydrocarbons etc [6]. Fly ash produced from thermal power plants has SiO<sub>2</sub>, Fe<sub>3</sub>O<sub>4</sub>, Fe<sub>2</sub>O<sub>3</sub>, TiO<sub>2</sub> etc as chief constituents. Moreover partially burnt carbon particles and a small fraction of Na, K, Mg, Zn, Pb, Cr, Cd, Ni, Fe, As, and Co in trace quantities are also available in it. These toxic trace elements acts as main source of pollution for soil, water (both surface and ground). Ground water is susceptible to metal contamination due to waste disposal of fly ash and leachate infiltration. The collection of huge quantity of coal fly ash in dumping location causes leakage of perilous substances and seepage through layers of soil and finally pollutes the ground water [7].

It thus becomes essential to recurrently keep an eye on the groundwater quality and to find out the ways to shield it from contamination. One very important and effective tool to get information of water quality is water quality index(WQI). First and foremost, the water quality index was introduced by Horton in U.S by choosing 10 most generally used parameters of quality of water like DO, pH, TDS, EC, etc [8]. Afterwards a number of methods has been proposed like weighted arithmetic water quality index, NSFQWI, CCMEWQI, etc. have been applied to estimate quality of water. Kaushik *et.al* (2002) [9] employed a total of nine parameters like electrical conductivity, pH etc. for determining water quality index for ground water of Hisar and Panipat using weighted arithmetic water quality index method and found that the WQI is less than 50 in Panipat region, which indicates that water is not fit for consumption whereas, it ranged between 50 and 100 for Hisar indicating poor water quality. Sunita *et.al* (2014) [10] employed weighted arithmetic water quality index

method for estimating the ground water quality of Samalkha, Panipat using ten parameters and reported that WQI lies in the range of 89.09 to 146.67 thus indicating poor water quality and not fit for drinking purposes. Khawaja *et .al.* (2016) [11] estimated water quality index for ground water samples of Aligarh, UP by incorporating Tiwari and Mishra method [12] using 14 parameters and found that WQI for this region ranged from 18.91 to 74.66 pre-monsoon and 16.81 to 70.35 during post-monsoon. The study revealed that half of the area chosen for research was moderately polluted. In the light of study of literature, it is obvious that several studies has been conducted to characterise physico-chemical parameters of ground water samples and estimate water quality index so far. However no study has been performed in Yamuna Nagar, Haryana, to characterise physico-chemical parameters and estimate the water quality in this region so far. Yamuna Nagar district is core industrial hub and is densely populated due to increasing number of immigrants from other states for exploring employment. The city has Sugar industry like sugar Mill and heavy machinery production like ISGEC and paper and pulp industry like Ballarpur paper and pulp industry and in the centre of the city, Deenbandhu thermal power plant. Thus from environmental pollution prone area point of view, Yamuna Nagar is best suited for our study.

## 2. Study Area

In our study, various samples of ground water were collected from tube wells situated in four villages and one residential colony surrounding the thermal power plant, Yamuna Nagar. Table 1 shows the location of sampling areas and Fig. 1 shows the geographical map of the sampling locations.

Table 1: Sampling locations surrounding thermal power plant Yamuna Nagar

Sr. No.	Sampling Code	Sampling Location	Source of ground water	Distance from Thermal Plant
1	TPYN-1	Rattan Pura	Tubewell	1 Km.
2	TPYN-2	Ram Pur	Tubewell	1.5 Km
3	TPYN-3	IshharPur	Tubewell	2 Km
4	TPYN-4	Bhadimajra	Tubewell	4 Km
5	TPYN-5	Power Plant Colony, Yamuna Nagar	Tubewell	0.5 Km



Fig.1: Geographical map of sampling site locations in the vicinity of thermal power plant Yamuna Nagar

## 3. Materials and Methods

The water samples from sampling locations of Yamuna Nagar were collected in sterilised plastic bottles. Temperature measurement was done in-situ using mercury thermometer and rest parameters are measured in the laboratory by adopting standard instruments and methods for examination of water and waste water according to American Public Health

Association manual (APHA 2005) and are given in table 2.

All the instruments stated in table 2 are calibrated against the standards. Table 3 shows the standard values of different parameters measured for characterization of water quality and have been taken from standards manuals of BIS and WHO.

Table 2: Details of Instruments used to measure various parameters

Sr. No.	Parameter	Instrument
1	pH	Eutech pH 510
2	Electrical Conductivity	Eutech CON 700
3	Total Dissolved Solids (TDS)	Cyberscan CON 1500
4	Total hardness	EDTA Titration
5	Turbidity	Eutech TN 100
6	Zn, Fe, Cu, Mn, Pb, Cd, Cr, Ni, As, Co, Ca, Mg	ICPE 9000, Shimadzu
7	Na, K	AAS, ZEE nit 700p, Analytic Zena

Table 3: Standard values of parameters of drinking water quality

These standard values of different parameters are widely used during the calculations of WQI. A number of methods have been devised for the WQI calculations after Horton method (1970). One such method is Tiwari and Mishra method, used in the present work for the calculation of WQI.

### 3.1 Tiwari & Mishra Water Quality Index

In this method, WQI is computed by the following equation:

$$WQI = \text{Antilog} \left[ \sum w_i \log_{10} q_i \right]$$

where  $w_i = \frac{K}{v_s}$

and  $K = \text{constant} = \frac{1}{(1/v_{s1} + 1/v_{s2} + \dots + 1/v_{sn})}$

where

$v_s =$

standard value of *i*th parameter  $q_i$  is calculated using the following equation:

$$q_i = \frac{(v_0 - v_i)}{(v_s - v_i)} \times 100$$

where

$v_0 =$  observed value of *i*th parameter

Table 5: Measured values of different parameters in vicinity of TPP Yamuna Nagar in Dec 2017

Measured parameter	Unit	Results (mean±s.d.)				
		TPYN-1	TPYN-2	TPYN-3	TPYN-4	TPYN-5
pH	-----	7.315±0.046	7.38±0.018	7.66±0.057	7.58±0.13	7.54±0.23
Electrical Conductivity	µS cm	1194.5±4.2	1249.7±30.6	1220.7±9.5	857± 16.0	1256.5±38
Total Dissolved Solids	ppm	832±11.13	861.2 ±13.14	831.5±11.5	573±10.7	855.2±19.7
Total hardness	ppm	353.75±19.5	326.75±11.7	328.25±16.9	284±8.0	362±15.0
Turbidity	NTU	5.63±0.10	5.47±0.08	5.10±0.09	4.74±0.17	5.62±0.10

Measured parameter	Standard value as per WHO(2017)	Standard value as per BIS(10500:2012)
pH	No guideline available in WHO manual	6.5- 8.5
Electrical Conductivity	<2500 µS cm	No guideline
Total Dissolved Solids	No guideline	500-2000 ppm
Total hardness	No guideline	200-600 ppm
Turbidity	No guideline	1-5
Zn	3 ppm	5-15 ppm
Fe	No guideline	<0.3 ppm
Cu	<2 ppm	0.05-1.5 ppm
Mn	0.5 ppm	0.1-0.3 ppm
Pb	<0.01 ppm	<0.01 ppm
Cd	<0.003 ppm	<0.003 ppm
Cr	<0.05 ppm	<0.05 ppm
Ni	<0.02 ppm	<0.02 ppm
As	<0.01 ppm	<0.01 ppm
Co	No guideline	No guideline
Na	200 ppm	No guideline
K	No guideline	No guideline
Ca	No guideline	75-200 ppm
Mg	No guideline	30-100 ppm

$v_i =$  ideal value of parameters (for pH,  $v_i = 7$  and for all other parameters,  $v_i = 0$ )

$v_s =$  standard value of *i*th parameter

As per Tiwari & Mishra method the rating scales for the quality of water are shown in Table 4.

Table 4. Rating Scales for Water Quality index as per Tiwari & Mishra method

WQI	Rank
0-25	Excellent quality
26-50	Good quality
51-75	Poor quality
76-100	Very Poor quality
Above 100	Unfit for drinking purposes

### 4. Results and Discussions

Physico-chemical parameters like, Electrical conductivity (EC), pH, TDS, Total hardness (TH), Turbidity, temperature and metal contents like Zn, Fe, Cu, Mn, Toxic trace elements Pb, Cd, Ni, Cr, As, Co and alkali contaminants Na, K, Ca and Mg are measured and compared with WHO guidelines (2017) [14] and BIS guidelines (10500: 2012) [15] standards. Table 5 and 6 shows the measured values in Dec 2017 (winter) and June 2018 (summer).

Temperature	°C	13.75±0.5	12.75±0.95	13±0.81	13±0.81	13.5±0.57
Zn	ppm	5.82±0.44	6.03±0.048	6.13±0.106	4.15±0.13	10.11±0.08
Fe	ppm	0.38±0.05	0.34±0.02	0.43±0.008	0.28±0.006	0.83±0.06
Cu	ppm	0.079±0.001	0.092±0.006	0.089±0.004	0.046±0.002	0.094±0.003
Mn	ppm	0.112±0.01	0.182±0.007	0.202±0.006	0.096±0.002	0.26±0.008
Pb	ppm	0.021±0.009	0.016±0.007	0.017±0.006	0.008±0.0006	0.033±0.005
Cd	ppm	0.005±0.0006	0.005±0.0008	0.005±0.0008	0.0018±0.0005	0.0038±0.0005
Cr	ppm	0.055±0.003	0.053±0.003	0.055±0.003	0.038±0.008	0.055±0.002
Ni	ppm	0.025±0.002	0.024±0.003	0.022±0.002	0.018±0.0008	0.024±0.002
As	ppm	0.014±0.003	0.014±0.002	0.014±0.003	0.008±0.001	0.013±0.002
Co	ppm	0.0038±0.0005	0.0028±0.001	0.0038±0.001	0.0013±0.0005	0.002±0.0005
Na	ppm	119.6±8.34	120.55±5.50	112.35±2.40	138.22±6.69	170.67±13.12
K	ppm	11.65±0.91	12.08±0.39	11.23±0.81	12.53±0.49	15.81±1.01
Ca	ppm	82.45±2.89	73.87±2.26	76.2±4.14	72.75±1.98	84.75±2.51
Mg	ppm	35.1±3.22	34.02±3.16	32.62±2.24	26.75±0.85	35.7±2.53

Table 6: Measured values of different parameters in vicinity of TPP Yamuna Nagar in June 2018

Measured parameter	Unit	Results (mean±s.d.)				
		TPYN-1	TPYN-2	TPYN-3	TPYN-4	TPYN-5
pH	-----	7.89±0.11	7.67±0.09	7.78±0.087	8.34±0.13	8.22±0.29
Electrical Conductivity	µS cm	1207±20.11	1242.75±15.4	1235.25±30	867±12.8	1282.25±11.2
Total Dissolved Solids	ppm	843±8.04	867±3.55	827.75±12.5	577.5±5.8	860.25±12.09
Total hardness	ppm	356±13.2	337±2.94	334±5.77	299.75±4.34	382.25±9.28
Turbidity	NTU	5.71±0.06	5.28±0.08	5.83±0.10	4.05±0.03	5.07±0.07
Temperature	°C	36.75±0.95	35.75±0.95	35.75±0.95	35.75±0.95	34.75±0.5
Zn	ppm	6.20±0.06	6.105±0.10	6.17±0.10	4.53±0.33	10.42±0.32
Fe	ppm	0.41±0.03	0.37±0.04	0.47±0.02	0.27±0.01	0.79±0.01
Cu	ppm	0.08±0.006	0.090±0.004	0.091±0.005	0.043±0.003	0.097±0.001
Mn	ppm	0.118±0.006	0.18±0.005	0.21±0.005	0.094±0.003	0.27±0.01
Pb	ppm	0.024±0.008	0.018±0.003	0.020±0.006	0.007±0.001	0.032±0.002
Cd	ppm	0.005±0.0005	0.005±0.0008	0.004±0.001	0.002±0.0008	0.003±0.0005
Cr	ppm	0.053±0.001	0.054±0.003	0.053±0.002	0.041±0.009	0.054±0.001
Ni	ppm	0.026±0.002	0.026±0.001	0.024±0.001	0.017±0.001	0.024±0.0008
As	ppm	0.015±0.001	0.016±0.001	0.015±0.001	0.008±0.0008	0.015±0.001
Co	ppm	0.003±0.0005	0.003±0.0008	0.004±0.0008	0.001±0.0006	0.003±0.0006
Na	ppm	122.3±8.77	122.0±3.11	116.0±1.32	139.0±3.84	171.5±10.5
K	ppm	11.53±0.67	11.66±0.27	11.14±0.40	12.07±0.08	15.43±0.57
Ca	ppm	82.0±3.8	78.8±0.83	75.95±1.13	72.6±0.94	85.97±1.44
Mg	ppm	35.8±1.11	33.1±1.17	33.9±0.76	28.0±0.88	37±0.90

There is a significant difference between the values of pH during two seasons i.e. summer and winter, the observed values lies in the interval 7.31 (TPYN-1) to 7.66 (TPYN-3) for the winter, i.e., Dec 2017 samples and between 7.67 (TPYN-2) to 8.34 (TPYN-4) for the summer, i.e., June 2018 samples. In the summer season, pH values are higher than those in winter season which reveals the dependence of pH on temperature. W. F. Langelier (1946) [16] and Castells C.B. et.al. (2003) [17] suggests a possible reason for this temperature dependence. As the temperature rises, molecular vibrations increase which results in the ability of water to ionise and form more hydrogen ions. As a result, the pH will drop. However in our study pH has higher values in summer. One of the possible cause for this anomaly is measurement of pH in laboratory environment whereas it must be done in-situ so that temperature of environment does not affect the measurement of pH. The values of electrical conductivity lies in range 857 µS cm (TPYN-4) to 1256.5 µS cm (TPYN-5) in winter and 867 µS cm (TPYN-4) to 1282.25 µS cm (TPYN-5) in summer.

Electrical conductivity does not show any variation with temperature. The presence of large amount of inorganic elements dissolved in water is obvious from the values of electrical conductivity [18]. These lie well in the range of normal electrical conductivity values as prescribed by WHO. The values of total dissolved solids are in the range 573 ppm (TPYN-4) to 861.2 ppm (TPYN-2) in winter season and 577.5 ppm to 867 ppm in the summer season. The hardness of water do not alter significantly with season [19]. Its value measured in winter season varies from 284 ppm to 353.75 ppm and 299.75 ppm to 382.25 ppm in the summer season and lie well in the range prescribed for normal value of hardness by BIS. As far as turbidity is concerned, there is very small deviation in the values from winter to summer. Concentration of various toxic pollutants like Pb, Cd, Cr, Ni, As, Co are having deviations from their standard values which indicates the pollution in ground water in the surrounding areas of thermal power plant [20]. The possible reason behind this contamination may be effluents discharges from thermal power plant [21-22].

Table 7 shows the values of water quality index (WQI) measured for the samples collected in summer and winter season both for different sampling locations. In

total 20 parameters are used to calculate WQI values using Tiwari & Mishra method.

Table 7: WQI values for samples collected in Dec 2017 (winter) and June 2018 (summer)

Sampling location Code	WQI value for samples collected in Dec 2017 (winter)	Water quality	WQI value for samples collected in June 2018 (summer )	Water quality
TPYN-1	271.64	Not fit	238.39	Not fit
TPYN-2	222.33	Not fit	236.10	Not fit
TPYN-3	268.53	Not fit	269.33	Not fit
TPYN-4	102.04	Not fit	87.94	Very poor
TPYN-5	246.55	Not fit	224.38	Not fit

The values of water quality index as shown in table 7 for sampling location TPYN-1 to TPYN-3 and TPYN-5 are very much above the quality rating for “not fit” water category. Moreover for these three sampling locations WQI values are not much altered by seasonal variation. For sampling location TPYN-4, i.e., Bhaadhimajra, WQI are 102.4 in winter and is 87.94 in summer. The big difference in WQI at this location from all other location is probably due to more distance between the thermal power plant and the village Bhaadhimajra and thus less prone to pollution [23-24].

### 5. Conclusions

The physico-chemical parameters for deciding the quality of ground water in the vicinity of thermal power plant, Yamuna Nagar are much higher than the standard values decided by WHO (2017) and BIS (10500:2012), and thus the water quality index values are much higher than that of normal water fit for drinking purposes. This is affecting not only the present human population of surrounding areas but may cause serious implications for the survival of flora and fauna in next few years. Moreover the values of various parameters measured in Dec 2017 i.e., winter and June 2018 i.e., summer undergoes mild seasonal variation.

### 6. Acknowledgement

The authors are thankful to Director, CSSRI, Karnal and Dean, RIC, I.K.G. Punjab Technical University, Kapurthala to provide necessary instrumentation to accomplish our work.

### 7. References

[1] Prasad N.R. and Patil J.M. A study of physico-chemical parameters of Krishna River water particularly in Western Maharashtra. *Rasayan J. Chem.* 1 (4): 943-958 (2008).

[2] Joseph R. and Tessy P.P. Water quality and pollution status of Chalakudy River at Kathikudam, Thrissur District, Kerala, India. *Nature Environment and pollution Technology*, 9 (1): 113-118 (2010).

[3] Hacıoglu N. and Dulger B. Monthly variation of some physico-chemical and microbiological parameters in Biga Stream ( Biga, Canakkale, Turkey). *African Journal of Biotechnology*, vol. 8 (9):1929-1937 (2009).

[4] Najafpour A.F.M., Alkari, Kadir M.O.A. and Najafpour Gh.D. Evaluation of spatial and Temporal variation in river water quality. *Int. J. Environ. Res.* 2 (4): 349-358 (2008).

[5] Rajeshwari C.V. and Saraswathi B. Assessment of water quality of Rivers Tungabhadra & Hundri, India. *Pollution Research* 28 (3): 499-505 (2009).

[6] Gasim M.B., Mir S.I. and Chek. T.C. A physico-chemical assessment of the Bebar river, Pahang, Malaysia. *Global Journal of Environmental Research*, 1 (1): 07-11 (2007).

[7] Lokhande R.S., Shinde D.N., Kulkarni S.W., Lohani P., Ghodvande V. and Gangele S. Hydrobiological studies of Ulhas River, Thane District (M.S.), India at various stations. *Pollution Research*. 27 (4): 735-738 (2008).

[8] Samantray P., Mishra B.K., Panda C.R. and Rout S.P. Assessment of Water Quality Index in Mahaadi and Atharabanki Rivers and Taldanda Canal in Paradip Area, India. *Journal of Human Ecology*, 26 (3):153-161 (2009).

[9] Kaushik A., Kumar K., Kanchan, Taruna, Sharma H.R. Water quality index and suitability assessment of urban ground water of Hisar and Panipat in Haryana. *Journal of environmental biology*, Jul;23 (3):325-33 (2002).

[10] K. Sunita and R. Jyoti. Assessment of water quality index of ground water in Smalkhan, Haryana, *International Journal of Latest Research in Science and Technology*, Volume 3 (6): Page No.169-172 (2014).

[11] Khwaja M.A., Bhattacharya G.S. and A. Vanita. Analysis of groundwater quality of Aligarh city, U.P.(India). *International Journal of Advancement in Engineering Technology, Management and Applied Science* vol3 (9) pp.41-48 (2016).

[12] Tiwari, T.N. and Mishra, M.A. A Preliminary Assignment of Water Quality Index of Major Indian River. *Indian Journal of Environmental Protection*, 5, 276-279 (1985).

[13] APHA (American Public Health Association) Standard method for examination of water and wastewater, NW, DC 20036 (1994).

- [14] Guidelines for drinking water quality-WHO, Geneva, (2) ED, 97-100 (2017).
- [15] Indian Standards, Bureau of Indian Standards (BIS), Indian standard specification for drinking water, IS – 10500, 2-4 (2012).
- [16] Langelier W.F. Effect of temperature on the pH of natural waters. *Journal of American water works association*. 38: 179 (1946).
- [17] Castells C.B., Ràfols C., Rosés M. and Bosch E. Effect of temperature on pH measurements and acid-base equilibria in methanol-water mixtures. *Journal of applied chromatography*, Jun 20;1002 (1-2): 41-53 (2003).
- [18] Razak A.A., Asiedu A.B., Entsua-Mensah R.E.M. and de Graft Johnson K.A.A. Assessment of water quality of the Oti river in Ghana. *West African J. of Applied Ecology*, Vol.15 (2009).
- [19] Kulandaivel A.R.K., Kumar P.E., Perumal V. and Magudeswaran P.N. Water Quality Index of River Bhavani at Erode Region, Tamil Nadu, India. *Nature, Environment and Pollution Technology*, 8 (3): 551-554 (2009).
- [20] Sonawane S.M. Effect of Waste Water and Ash of Thermal Power Station on Hydrological Features of River Tapi at Bhusawal, District Jalgaon Maharashtra. *IOSR Journal of Environmental Science, Toxicology and Food Technology*, Volume 9, Issue 7, Ver. II, PP 19-24 (2015).
- [21] Mohapatra P.D., Patra B.B. and Kar B.B. Physico chemical analysis of ground water near a thermal power plant. *International Journal of Innovative Research in Science, Engineering and Technology*, vol 2, issue 7 (2013).
- [22] Hasan I., Rajia S., Kasi A.K. and Latifa G.A. Comparative study on the water quality parameters in two rural and urban rivers emphasizing on the pollution level. *Global Journal of Environmental Research* 3 (3): 218-222 (2009).
- [23] Jadav R.K., Saini S.S. and Kallai S. Environmental Assessment of Underground Water Quality near Suntgarh Super Thermal Power Plant in Sriganganagar District, Rajasthan. *The Eco, Canada*, Vol. 4, 347-349 (2010).
- [24] Mandal A. and Sengupta D. Radionuclide and Trace Element Contamination around Kolaghat TPS, WB, India-Possible Environmental Hazards. *Environmental Geology*, Vol. 44, 180-187 (2005).