

Antioxidant Activity and Total Phenol Anthocyanin and Flavonoid Content Analysis of Raspberry

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Abstract

This study aim to investigate phytochemical and antioxidant activity of *Rubus idaeus* L fruit. The total phenolics, flavonoids and anthocyanins were evaluated. The antioxidant activities were investigated using three antioxidant assays 2,2-diphenyl-1-picrylhydrazyl (DPPH), Ferric Reducing (FRAP) and 2,2'-azinobis(3-ethylbenzothiazoline-6-sulfonic acid) (ABTS). The raspberry fruit contain of total phenolics (37.19±0.39 mg GAE/g), flavonoids (5.32±0.13 mg RU/ g) and total anthocyanin content (32.21±0.35 mg c-3-gE/g) were recorded. The antioxidant activity in DPPH assay IC₅₀ value 39.00±1.32 µg/ml, ABTS assay (0.85±0.02 mg AEAC/g) and FRAP assay 52.03±4.65 mM Fe²⁺/g were evaluated. These results presenting perspectives usage of *R. idaeus* fresh fruits was considerable levels of phytochemicals and antioxidant activity.

Keywords *Phytochemical, Antioxidant, Rubus idaeus, 2,2-diphenyl-1-picrylhydrazyl, phenolics.1.*

1. Introduction

The *Rubus idaeus* L. (raspberry) is a exclusive berry with a rich history. Consuming a healthy and diet loaded in vegetables and fruits is related with a compact risk of some lifestyle and noncommunicable age related diseases such as cancer, type2 diabetesmellitus (T2DM), cardiovascular disease (CVD) and Alzheimer disease. Red raspberries supply to the relating to diet value. It is food sources of dietary fiber 6.5 g/100 g and calorie 12.5 g/100 kcal. It also have magnesium, vitamin C, potassium, vitamin K, calcium, and iron (USDA. Scientific report 2015).

Raspberry polyphenols principally consist of hydrolysable tannins and anthocyanins. Particularly, raspberries are a abundant source of cyanidin glycosides as well as unique from other berries due to high ellagitannin content resulting in the discharge of free ellagic acid during hydrolysis (Rao, A.V et al., 2010). Ellagitannins were showed extensively add to the antioxidant activity of raspberries.

Ellagitannins were responsible for 58% of antioxidant ability of raspberry fruits (Borges, G. et al., 2010). In addition, ellagitannins are exacting interest for nutritional and pharmacological industries because of compounds may act as effective chemopreventive agents (Lande, J.M 2011).

In additionally minerals and vitamins, the phytochemicals in raspberry has been related to such kind of benefits reduced risk of cardiovascular disease, weight management, reduced blood pressure and cholesterol, improved cognitive brain function, stroke and slowed agerelated eyesight degeneration (Zafra-Stone et al., 2007). Bioactive compounds can work as proteins, antioxidants, lipids, stopping or limiting damage to cellular DNA caused by reactive oxygen species. A strong relation between eating of antioxidant loaded foods and cancers and decreased risk of cardiovascular disease (Heinonen, I et al., 1998).

Raspberries extracts have been established to make use of antimicrobial, anti-inflammatory, antioxidant, anticancer, anti-Alzheimer activities and anthelmintic (. Jim'enez-Arellanes et al., 2012; C. S. Bowen-Forbes et al., 2010; H. Jung et al., 1886) . The bioactivities of raspberry are mostly due to the presence of phytochemicals. Previous studied have shown that black raspberries presented the highest amount of anthocyanin, total polyphenols and flavonoid while compared to Korean raspberries and blackberries. It is also reported that anti-inflammatory activities and high antioxidant activity by raspberry due to high concentration of polyphenols and anthocyanin (Heinonen, I et al., 1998).

2. Materials and Methods

2.1 Materials

Raspberry fruits were procured by local medicinal herbs supplier. Fruits were air dried and made it powder and stored in laboratory condition.

2.2 Reagents and Solvents

All reagents and solvents 2,2-diphenyl-1-picrylhydrazyl (DPPH free radical) and diphenylboryloxyethylenamine (DPhBOA) were procured from Sigma-Aldrich.

2.3 Extract Preparation

Raspberry powder 20 g was extracted with 100 ml methanol at room temperature for 24 hrs under constant shaking. The extract were filtered and dried in a rotary vacuum evaporator to remove methanol. After evaporation, the residue was suspended in methanol for further analysis.

2.4 Total Anthocyanin Content.

Total anthocyanin content was determined by with a spectrophotometric pH differential method with some modification (M. M. Giusti et al., 2001). The reaction mixture follow as, 0.5mL extract was added 3.5mL of potassium chloride buffer (0.025M, pH 1.0). The mixture was mixed and kept to stand for 15 minutes. The absorbance were calculated at 515 and 700 nm. distilled water used as blank. Results were presented as mg cyanidin-3-glucoside equivalents (c-3-gE)/g of dried sample.

The total anthocyanins content was calculated by following formula:

Total anthocyanin content (mg/g of dried sample)

$$\frac{A \times Mw \times DF \times 10}{(\epsilon \times C)}$$

where A is absorbance = $(A_{515} - A_{700})$, pH 1.0 – $(A_{515} - A_{700})$

pH 4.5, Mw is molecular weight for cyanidin-3-glucoside = 449.2, DF is a dilution factor of the samples, ϵ = the molar absorptivity of cyanidin-3-glucoside = 26,900, C = the concentration of the buffer in mg/mL.

2.5 Total phenolic content

The Total Phenol Content was investigated using Folin-Ciocalteu method by Wootton-Beard (2011) with some modifications. 0.5 mL of extract mixed with (10%) 2.5 mL Folin-Ciocalteu reagent and 2.5 mL NaHCO_3 (7.5%). The reaction mixture was incubated at 45 °C for 15 min. Blank was prepared in the same way without adding of extract. The absorbance of were measured on the spectrophotometer at 765 nm. Results were presented as mg gallic acid equivalents of dried sample.

2.6 Total flavonoid content

The total flavonoids content was determined by aluminum chloride (Brighente IMC et al., 2007). Briefly, 1 mL of the extract was mixed with (2%) 1 mL of AlCl_3 . The reaction mixture was incubated for 1 hour at room temperature. The absorbance was

observed at 415 nm. Total flavonoid content were expressed as mg of RU g⁻¹.

2.7 DPPH free radical scavenging assay

DPPH free radical scavenging activity of raspberry extract was determined using DPPH method (Takao T et al., 1994). 1 mL of DPPH solution ((2 mg mL⁻¹), 1 mL of different concentrations of the extract or the standard solution was added independently. The reaction mixtures were incubated at 37 °C for 30 min. The absorbance were measured at 517 nm using methanol as blank. The DPPH scavenging activity (%) of extract and standard AA was determined using the following equation:

$$\% \text{inhibition} = [(Ac - As) / Ac] \times 100$$

Where Ac = absorbance of control reaction, As = absorbance of the sample.

2.8 ABTS decolorization assay

The ABTS·⁺ radical cation decolorization assay was performed spectrophotometric method. The ABTS⁺ scavenging activity was calculated previously described by Jakovljević (Jakovljević VD et al., 2016). Briefly, ABTS stock solution (7 mM) with 2.45 mM potassium persulfate. The mixture was incubated in dark condition at room temperature for 16 hrs. Under this condition, ABTS⁺ could be stable in this form for 2 days. The ABTS⁺ solution was diluted double distilled water to obtain an absorbance of 0.70±0.02 at 734nm. Aliquots of 30 μL of the extract different concentrations then added to 2.7 mL diluted ABTS·⁺ solution, and mixture was incubated at room temperature for 30 min. Absorbance was recorded at 734 nm. AA, was used as standards. The percentage of inhibition was calculated using foloowing formula:

$$\% \text{inhibition} = [(Ac - As) / Ac] \times 100$$

Where Ac = absorbance of control reaction, As = absorbance of the sample.

2.9 Ferric ReducingAntioxidant Power Assay (FRAP)

The capability of the extract to reduce ferric ion (Fe^{3+}) to ferrous ion (Fe^{2+}) was determined using previous method I. F. F. Benzie and J. J. Strain (I. F. F. Benzie and J. J. Strain 1996) with some modification. Briefly, 300mM acetate buffer (pH 3.6) was mixed with 10mM TPTZ and 20mM $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ with 10:1:1 ratio. FRAP reagent was used as a blank and measured at 593 nm. 100 μL extract and 300 μL distilled water were added to the blank. Took second reading after 4 minutes. $\text{Fe}(\text{II})$ was prepared as a standard with numerous concentrations from 0 to 100μg/mL. The results were shown as the concentration of antioxidant having a ferric reducing ability in 1 gram of sample (mM/g).

3. Results and Discussion

3.1 Total Phenolic, Flavonoid and Anthocyanin Contents

Anthocyanins, Phenolics and Flavonoids are the phytochemicals that usually presented in berries and known to have antioxidant, anticancer, anti-inflammatory, antimutagenic, antihypertension and antineurodegenerative. Naturally, these phytochemicals are very essential components for physiological functions of plants such as pathogens, for pollination, herbivore and protection against UV light^[9]. Therefore, the amount of the total phenolic, anthocyanin and flavonoid content *Rubus idaeus* was determined. The results were presented in Table 1. The raspberry fruit extract phenol content 37.19±0.39 mg GAE/g, flavonoid content 5.32±0.13 mg RU/ g were recorded. Total anthocyanin contents 32.21±0.35 mg c-3-gE/g was displayed by *Rubus idaeus* fruit. or above the table as specified using font size of 9 or 10.

Table 1. Anthocyanin, Phenol and Flavonoid Content of *Rubus idaeus* fruit.

	TPC (mg GA/g)	TAC (c-3-gE)/g DS	TFC (mg RU/ g)
Methanol Extract	37.19±0.39	32.21±0.35	5.32±0.13

3.2 DPPH, FRAP and ABTS Assays

The antioxidant capacities of *Rubus idaeus* was measured using three dissimilar *in vitro* antioxidant assays. In DPPH assay, DPPH solution's purple color changed into yellow color due to presence of antioxidant compound. The antioxidant capacity of extract on DPPH free radical was presence of hydrogen-donating ability. The IC₅₀ value 39.00±1.32 was measured of raspberry fruit.

FRAP method used to determine the antioxidant capacity. In this method, ferric ion is reduced to ferrous ion at low pH, colored ferrous tripyridyltriazine complex formed. The reducing ability of *Rubus idaeus* fruit 52.03±4.65 mM Fe²⁺/g was recorded.

ABTS decolorization assay is similar to DPPH assay, which is the scavenging activity of the free radicals. However, the ABTS salt generated by chemical and enzymatic reaction first. The *Rubus idaeus* fruit extract shown lower scavenging effects against ABTS radicals. The raspberry fruit was recorded 0.85±0.02 mg AEAC/g ABTS scavenging activity. FRAP, ABTS and DPPH assays results were presented in table 2.

Table 2. DPPH, ABTS and FRAP assays of *Rubus idaeus* Fruit

	DPPH	ABTS	FRAP
Methanol Extract	39.00±1.32	0.85±0.02	52.03±4.65

DPPH free radical scavenging activity represented by IC₅₀ was expressed as µg/mL.

ABTS free radical scavenging activity was expressed as mg ascorbic acid equivalent antioxidant capacity (AEAC) in 1 g of dry sample.

FRAP was expressed as mM ferric reduction to ferrous in 1 g of dry sample.

5. Conclusions

Raspberries are an excellent natural source of antioxidant compounds such as tannins. Raspberry rising consumer interest to antioxidant phenotypes in a healthy diet. The antioxidant phytochemicals, several anthocyanins and ascorbic acid are present in raspberry. Our results specify that inconsistency in phytochemicals content of *Rubus idaeus* investigated which may be due to the environmental and genetic factors. Methanol extract of *R. idaeus* fruit showed a major quantity of phytochemicals, which gives to antioxidant, antibacterial, and antiacetylcholinesterase activities. Consumption of *Rubus idaeus* fruit in diet can offer healthy benefit.

6. Reference

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